

**SYLLABUS**  
**Fall semester 2020-2021 academic years**  
**on the educational program “8D07104-Chemical technology of inorganic substances”**

Discipline's code	Discipline's title	Independent work of students (IWS)	No. of hours per week			Number of credits	Independent work of student with teacher (IWST)
			Lectures (L)	Practical training (PT)	Laboratory (Lab)		
NP 8303	Inorganic polymers	98	15	30	-	5	7
<b>Academic course information</b>							
Form of education	Type of course	Types of lectures		Types of practical training	Number of IWS	Form of final control	
Online	Elective/Theoretical	Informational, lecture-conference		Seminar conference	4	Exam	
Lecturer	Prof. Burkitbayev M.M., first vice-rector						
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Telephone number	+7(727)221-11-23						
<b>Academic presentation of the course</b>							
Aim of course	Expected Learning Outcomes (LO)			Indicators of LO achievement (ID)			
	As a result of studying the discipline the undergraduate will be able to:			(for each LO at least 2 indicators)			
Develop the ability to evaluate and optimize technologies for the synthesis of polymer materials based on phosphorus-containing compounds.	<b>LO 1</b> Explain the technology and methods of synthesis of inorganic polymers			<b>ID 1.1.</b> Demonstrate knowledge of the synthesis of inorganic polymeric compounds; <b>ID 1.2.</b> Know the main steps of the synthesis of inorganic polymeric compounds			
	<b>LO 2</b> Know the methods of Predicting and calculating conditions for preparation of inorganic polymers.			<b>ID 2.1.</b> Predict and calculate the necessary conditions for the synthesis of polymer compounds; <b>ID 2.2.</b> Create a simple scheme for the synthesis of a given inorganic polymer			
	<b>LO 3</b> Apply strategic analysis tools for choice raw materials for preparation of inorganic materials.			<b>ID 3.1.</b> Classify raw materials for production of inorganic polymeric compounds; <b>ID 3.2.</b> Evaluate mineral resources base of Kazakhstan for production of inorganic polymeric compounds			
	<b>LO 4</b> Form strategies for choice synthesis technology			<b>ID 4.1.</b> Justify the appropriateness of the chosen synthesis technology; <b>ID 4.2.</b> Be objective in analyzing the current situation of synthesis			
	<b>LO 5</b> Develop new methods of preparation of inorganic polymers			<b>ID 5.1.</b> Able to develop new technological schemes for obtaining inorganic nanomaterials <b>ID 5.1.</b> Make a financial calculation of developed technology.			
<b>Prerequisites</b>	Theoretical foundations of inorganic chemistry; Inorganic chemistry; Mineral Raw Materials of Kazakhstan. Waste-free technology. Technology of inorganic acids, bases and salts						
<b>Post requisites</b>	Implementation of PhD thesis						
<b>Information resources</b>	Educational literature: 1. Charles E. Carraher, Jr. Introduction to polymer chemistry. CRC Press Taylor&Francis Group. 2013 2. V.W. Willison and L. Peck. Experiments in General Chemistry: Inquiry and Skill Building. Brooks/Cole (2009). 3. B. Stanton, L. Zhu, C. Atwood. Experiments in General Chemistry: Featuring Measurement. Brooks/Cole (2010). Internet resources: 1. Cuidi Li, Li Gao, Fangping Chen, Changsheng Liu Fabrication of mesoporous calcium silicate/calcium phosphate cement scaffolds with high mechanical strength by freeform fabrication system with micro-droplet jetting// Journal of Materials Science. November 2015, Volume 50, Issue 22, pp 7182-7191						

	<p>2. Longgong Xia, Zhihong Liu, Pekka Antero Taskinen Experimental determination of the liquidus temperatures of the binary (SiO<sub>2</sub>-ZnO) system in equilibrium with air// Journal of the European Ceramic Society, Volume 35, Issue 14, November 2015, Pages 4005–4010</p> <p>3. Chia-Tze Kao, Chi-Chang Lin, Yi-Wen Chen, Chia-Hung Yeh, Hsin-Yuan Fang, Ming-You Shie Poly(dopamine) coating of 3D printed poly(lactic acid) scaffolds for bone tissue engineering// Materials Science and Engineering: C, Volume 56, 1 November 2015, Pages 165–173</p> <p>4. Ammar Z. Alshemary, Muhammed Akram, Yi-Fan Goh, Usman Tariq, Faheem K. Butt, Ahmad Abdolahi, Rafaqat Hussain Synthesis, characterization, in vitro bioactivity and antimicrobial activity of magnesium and nickel doped silicate hydroxyapatite// Ceramics International, Volume 41, Issue 9, Part B, November 2015, Pages 11886–11898</p> <p>5. Wilaiwan Leenakul, Prathana Intawin, Tawee Tunkasiri, Jetsada Ruangsuriya, Kamonpan Pengpat Preparation of ferrimagnetic BF based silicate glass system// Ceramics International, Volume 41, Supplement 1, July 2015, Pages S464– S470 Mohammad, Faruq; Arfin, Tanvir; Al-Lohedan, Hamad A. Synthesis, Characterization and Applications of Ethyl Cellulose-Based Polymeric Calcium(II) Hydrogen Phosphate Composite// JOURNAL OF ELECTRONIC MATERIALS 47 (5): 2954-2963 DOI: 10.1007/s11664-018-6118-8</p>
<b>Academic policy of the course in the context of university moral and ethical values</b>	<p><b>Academic Behavior Rules:</b> Compulsory attendance in the classroom, the impermissibility of late attendance. Without advance notice of absence and undue tardiness to the teacher is estimated at 0 points. Submission of assignments (Independent work of students, midterm control, laboratory tasks and etc.) prior to the deadlines. The violation of submission deadlines leads to the deduction of penalty points. The submitting of laboratory works is only 2 weeks after their implementation!!!</p> <p><b>Academic values:</b> - Practical trainings/laboratories, IWS should be independent, creative. - Plagiarism, forgery, cheating at all stages of control are unacceptable. - Students with disabilities can receive counseling at e-mail Mukhambetkali.Burkitbayev@kaznu.kz, phone +7(727)221-11-23</p>
<b>Evaluation and attestation policy</b>	<p><b>Criteria-based evaluation:</b> assessment of learning outcomes in relation to descriptors (verification of the formation of competencies in midterm control and exams).</p> <p><b>Summative evaluation:</b> assessment of work activity in an audience (at a webinar); assessment of the completed task.</p>

**CALENDAR (SCHEDULE) THE IMPLEMENTATION OF THE COURSE CONTENT:**

week s	Topic name	LO	ID	amount of hours	Maximum score	Form of Knowledge Assessment	The Form of the lesson / platform
1	<b>L.1</b> Organic, organometallic and inorganic polymers.	LO 1	ID 1.1.	1			Offline
	<b>PT 1</b> Determination of normal consistency of cement paste.	LO 1	ID 1.1.	2	15	Analysis, IW	Offline
2	<b>L.1</b> The spread of inorganic polymers in nature. The differences between the HMC and inorganic polymers.	LO 1	ID 1.1.	1			Offline
	<b>PT 1</b> Determination of normal consistency of cement paste.	LO 1	ID 1.1.	2	15	Analysis, IW	Offline
3	<b>L.1</b> Homo chain inorganic polymers	LO 1	ID 1.1.	1			Offline
	<b>PT 1</b> Determination of setting time of cement.	LO 1	ID 1.1.	2	15	Analysis, IW	Offline
	<b>IWSP 1 Consultation on the implementation of IWS1</b>	LO 1	ID 1.1.	1			Offline
	<b>IWS 1.</b> Classification of polymer substances.	LO 2	ID 2.1.		10	Logic task	
4	<b>L.1</b> Hetero chain inorganic polymers.	LO 3	ID 3.1.	1			Offline
	<b>PT 1</b> Determination of setting time of cement.	LO 1	ID 1.1.	2	15	Analysis, IW	Offline
5	<b>L.1</b> The polymeric compounds of elements of groups I and II of periodic system.	LO 4	ID 4.1.	1			Offline

	<b>PT 1</b> Determination soundness of cement.	LO 1	ID 1.1.	2	15	Analysis, IW	Offline
	<b>IWSP 2 Consultation on the implementation of IWS 2</b>	LO 1	ID 1.1.	1			Offline
	<b>IWS 2</b> Chemical bonding in polymer substances.	LO 4	ID 4.1.		15	Logic task	
	<b>Make a structural and logical diagram of the read material</b>	LO 1	ID 1.1.				
5	<b>MT 1</b>	LO 1	ID 1.1.		100		
6	<b>L.1</b> Polymeric compounds of elements of groups III and IV of the periodic system.	LO 1	ID 1.1.	1			Offline
	<b>PT 1</b> Determination of the mechanical properties of the cement. Bending strength	LO 1	ID 1.1.	2	15	Analysis, IW	Offline
7	<b>L.1</b> Polymer compound V elements and VI, VII, VIII of the periodic system.	LO 1	ID 1.1.	1			Offline
	<b>PT 1</b> Determination of the mechanical properties of the cement. Bending strength	LO 5	ID 5.1.	2	15	Analysis, IW	Offline
8	<b>L.1</b> Polymer aluminum compound.	LO 1	ID 1.1.	1			Offline
	<b>PT 1</b> Properties of gypsum. Determination of normal density of gypsum paste.	LO 1	ID 1.1.	2	15	Analysis, IW	Offline
	<b>IWSP 3 Consultation on the implementation of IWS3</b>	LO 1	ID 1.1.	1			Offline
	<b>IWS 3</b> Silicates	LO 1	ID 1.1.		35	Logic task	
9	<b>L.1</b> Polycondensation reactions	LO 1	ID 1.1.	1			Offline
	<b>PT 1</b> Determination of free silicon in cement	LO 1	ID 1.1.	2	15	Analysis, IW	Offline
10	<b>L.1</b> The polymeric silicon compound.	LO 1	ID 1.1.	1			Offline
	<b>PT 1</b> Determination of the free (ions) silicon in silica.	LO 1	ID 1.1.	2	15	Analysis, IW	Offline
	<b>IWSP 4 Discussion of ISW 3</b>	LO 1	ID 1.1.	1			Offline
	<b>IWSP 5 Make a structural and logical diagram of the read material</b>	LO 1	ID 1.1.	1			
10	<b>MT (Midterm Exam)</b>	LO 1	ID 1.1.		100		
11	<b>L.1</b> Oxysilicon acid.	LO 1	ID 1.1.	1			
	<b>PT 1</b> Preparation of nitric acid: titrimetric determination of the reaction yield.	LO 1	ID 1.1.	2	15	Analysis, IW	Offline
12	<b>L.1</b> orthosilicic acid. Polycondensation of orthosilicic acid.	LO 1	ID 1.1.	1			Offline
	<b>PT 1</b> Determination of free aluminum (aluminum ions) in the clay.	LO 1	ID 1.1.	2	15	Analysis, IW	Offline
	<b>IWSP 6 Consultation on the implementation of IWS4</b>	LO 1	ID 1.1.	1			Offline
	<b>IWS 4</b> Alumosilicates	LO 1	ID 1.1.		15	Problem task	
13	<b>L.1</b> The salts of orthosilicic acid (silicates).	LO 1	ID 1.1.	1			Offline
	<b>PT 1</b> Determination of free aluminum (aluminum ions) in the clay.	LO 1	ID 1.1.	2	15	Analysis, IW	Offline
14	<b>L.1</b> Aluminosilicate.	LO 1	ID 1.1.	1			Offline
	<b>PT 1</b> Determination of free aluminum (aluminum ions) in the clay.	LO 1	ID 1.1.	2	15	Analysis, IW	Offline
15	<b>L.1</b> The high-temperature synthesis of aluminosilicates.	LO 1	ID 1.1.	1			Offline
	<b>PT 1</b> Determination of free aluminum (aluminum ions) in the clay.	LO 1	ID 1.1.	2	15	Analysis, IW	Offline

	<b>IWSP 7 Submission of ISW 1-4</b>	LO 5	ID 5.1.	1	10	Problem task	Offline
	<b>MT 2</b>	LO 1	ID 1.1.		100		

[Abbreviations: QS - questions for self-examination; TK - typical tasks; IT - individual tasks; CW - control work; MT - midterm.

Comments:

- Form of L and PT: webinar in MS Teams / Zoom (presentation of video materials for 10-15 minutes, then its discussion / consolidation in the form of a discussion / problem solving / ...)
- Form of carrying out the CW: webinar (at the end of the course, the students pass screenshots of the work to the monitor, he/she sends them to the teacher) / test in the Moodle DLS.
- All course materials (L, QS, TK, IT, etc.) see here (see Literature and Resources, p. 6).
- Tasks for the next week open after each deadline.
- CW assignments are given by the teacher at the beginning of the webinar.]

**Dean**  
**Chairman of the Faculty Methodical Bureau**  
**Head of the Department**  
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